## **Chapter 05 Worksheet 01**

Draw the reaction energy diagram, and label the products, reactants, transition states, and intermediates. 1. 3-Step reaction 2<sup>nd</sup> step is Rate-Determining-Step 3<sup>rd</sup> step is faster than the 1<sup>st</sup> 1<sup>st</sup> step is endothermic 2<sup>nd</sup> step is exothermic 3<sup>rd</sup> step is exothermic The overall reaction is exothermic Reaction progress 2. 3-Step reaction 3<sup>rd</sup> step is Rate-Determining-Step 1<sup>st</sup> step is faster than the 2<sup>nd</sup> 1<sup>st</sup> and 2<sup>nd</sup> step are endothermic 3<sup>rd</sup> step is exothermic The overall reaction is neutral Reaction progress 3. 3-Step reaction 2<sup>nd</sup> step is faster than the 3<sup>rd</sup>
1<sup>st</sup> and 2<sup>nd</sup> step are endothermic 3<sup>rd</sup> step is exothermic The overall reaction is exothermic Reaction progress

4.

- 4-Step reaction
  3<sup>rd</sup> step is Rate-Determining Step
  1<sup>st</sup> step is endothermic
  1<sup>st</sup> and 2<sup>nd</sup> step proceed at the same speed
  3<sup>rd</sup> step is exothermic
  4<sup>th</sup> step is exothermic

- The overall reaction is exothermic

Reaction progress

5.

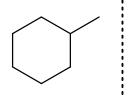
- - Reaction progress

- 3-Step reaction
- 3<sup>rd</sup> step is Rate-Determining-Step 1<sup>st</sup> step is faster than the 2<sup>nd</sup> 1<sup>st</sup> and 3<sup>rd</sup> steps are exothermic 2<sup>nd</sup> step is endothermic

- The overall reaction is endothermic

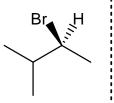
Draw mirror images of the following.

6.



7.

8.



9.

14.

$$H_3C$$
  $H_3C$   $H_2CH_3$ 

15

$$F$$
 $CH_3$ 
 $CH_3$ 
 $OH$ 

16.

$$F$$
 $CH_3$ 
 $CH_3$ 

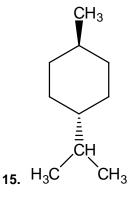
17.

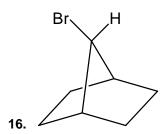
18.

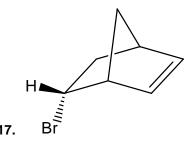
Label the chiral carbons with an asterisk (\*), and determine whether the compound itself is chiral or achiral.

$$_{f 9.}$$
 CH $_3$   $\stackrel{
m NH}_2$ 

13.







$$c = c$$

$$CH_3/I_{I_1}$$
  $C=C=C$